

so, $c = i\delta$

minimum uncertainty packet is:

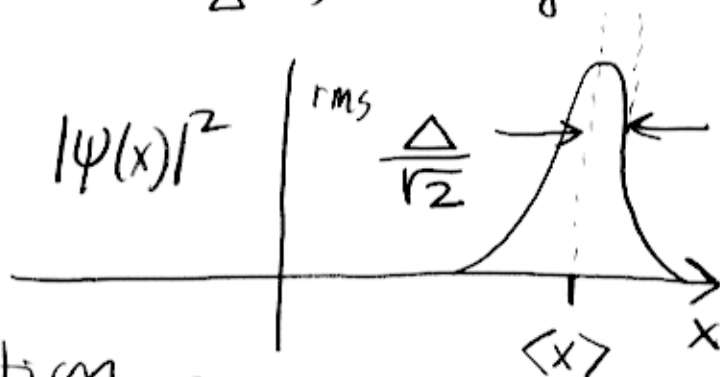
$$\Psi(x) = A e^{\frac{i\langle p \rangle x}{\hbar}} e^{-\frac{\gamma x'^2}{2\hbar}} \quad x' = x - \langle x \rangle$$

for $\Psi(x) \rightarrow 0$ as $x \rightarrow \pm\infty$, $\gamma > 0$

refer to p.135 of the text;

$$\frac{\gamma}{\hbar} = \frac{1}{\Delta^2} \quad \text{or} \quad \gamma = \frac{\hbar}{\Delta^2}, \quad \Delta^2 = \frac{\hbar}{\gamma}$$

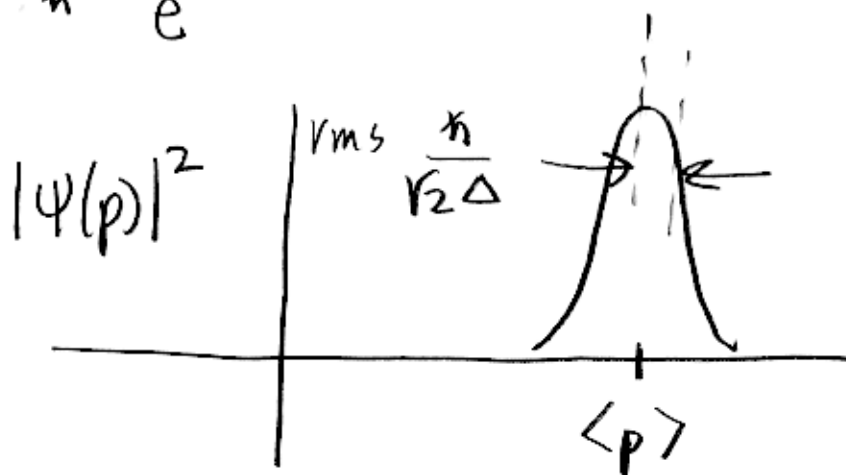
$$A = \frac{1}{(\pi \Delta^2)^{1/4}}$$



Δ quantifies the width of the distribution in space... Variance is $\Delta^2/2$.

$$\Psi(p) = \frac{1}{(2\pi\hbar)^{1/2}} \int_{-\infty}^{\infty} dx e^{-ipx/\hbar} \Psi(x) \quad (\text{p.137})$$

$$= \left(\frac{\Delta^2}{\pi\hbar^2}\right)^{1/4} e^{-\frac{i(p-\langle p \rangle)\langle x \rangle}{\hbar}} e^{-\frac{(p-\langle p \rangle)^2 \Delta^2}{2\hbar^2}}$$



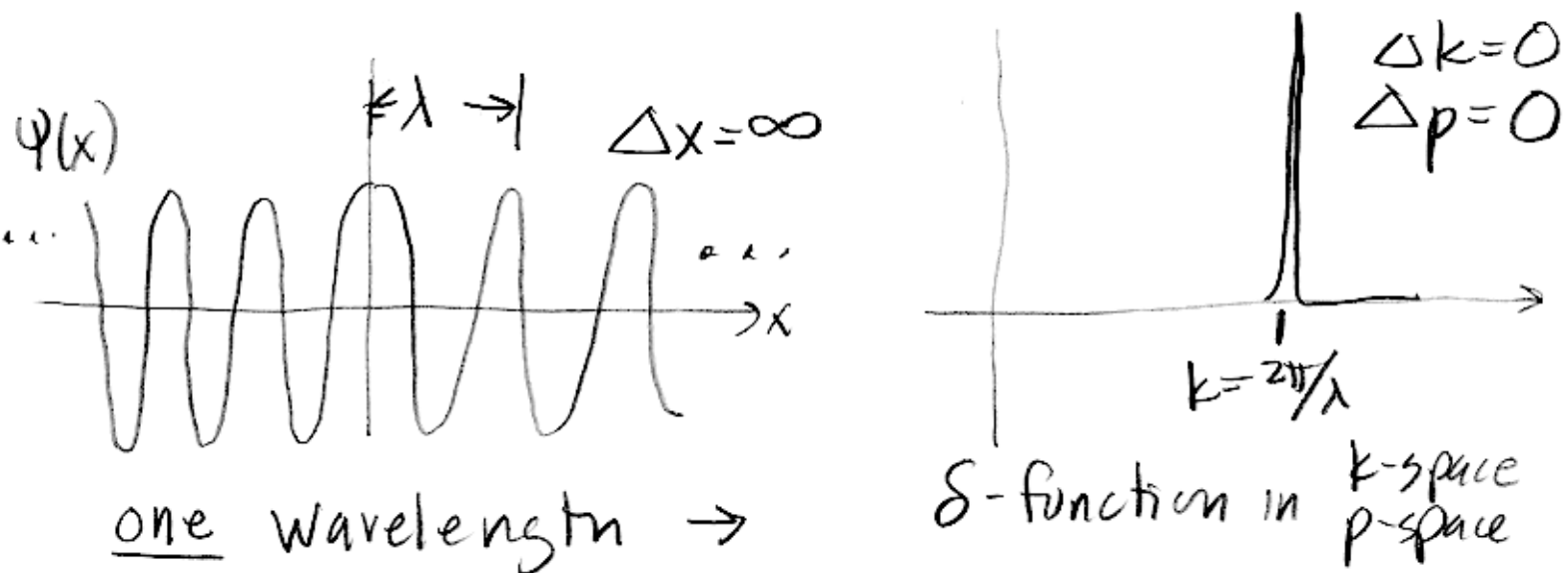
rms x \times rms p

$$\frac{\Delta}{\sqrt{2}} \times \frac{\hbar}{\sqrt{2}\Delta} = \frac{\hbar}{2}$$

A more qualitative view

uncertainty.xls = Excel Spreadsheet
available on course page

idea:



one wavelength \rightarrow

$$\psi \propto \cos\left(\frac{2\pi x}{\lambda}\right) = \operatorname{Re}\left(e^{\frac{2\pi i x}{\lambda}}\right) = \operatorname{Re}\left(e^{ikx}\right) = \operatorname{Re}\left(e^{\frac{ipx}{\hbar}}\right)$$

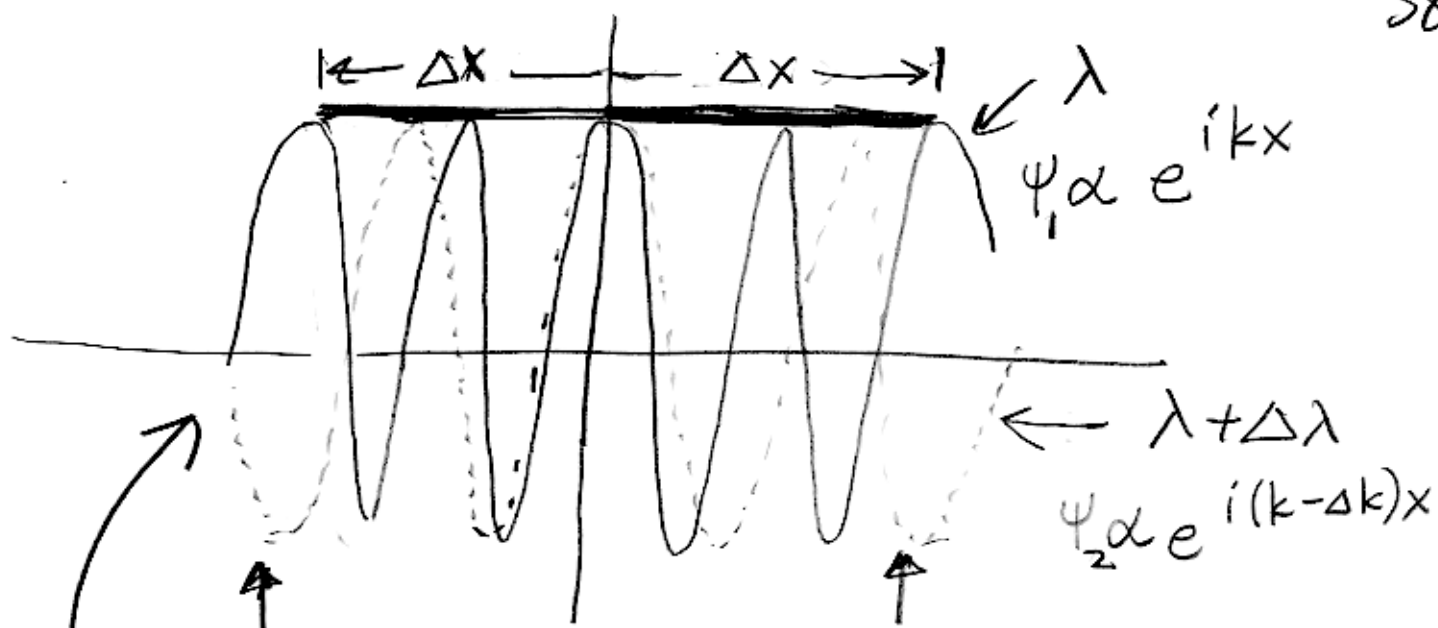
↑
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wavenumber $k = \frac{2\pi}{\lambda}$
 $p = \hbar k$

now, say you want to attempt to localize ψ in some region $2\Delta x$ wide.

Exploit the tool of superposition... blend in a wave with a slightly different wavelength $\lambda + \Delta\lambda$ and wavenumber $k - \Delta k$. How do you choose $\Delta\lambda$ or Δk ?

(recall, $\frac{2\pi}{\lambda + \Delta\lambda} \equiv k - \Delta k$)



Destructive interference at $\pm \Delta x$: $\psi_1 + \psi_2$ vanishes there.

$\psi_1 + \psi_2$ more localized in x-space

in k-space

$\psi_1 + \psi_2$ less localized in k-space.



want ψ_2 to have one fewer wavelength in the range $2\Delta x$, to get destructive interference

$$\frac{2\Delta x}{\lambda + \Delta\lambda} = \frac{2\Delta x}{\lambda} - 1 \quad (\text{Key Point})$$

or
$$\frac{2\pi}{\lambda + \Delta\lambda} \cdot \Delta x = \frac{2\pi}{\lambda} \cdot \Delta x - \pi$$

$$k - \Delta k \equiv \frac{2\pi}{\lambda + \Delta\lambda} \quad k = \frac{2\pi}{\lambda}$$

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$$(\cancel{k} - \Delta k) \cdot \Delta x = \cancel{k} \Delta x - \pi$$

$$\Delta k \Delta x = \pi$$

$$\text{or } \underbrace{\hbar \Delta k \Delta x}_{\Delta p} = \hbar \pi$$

$$\boxed{\Delta p \Delta x = \hbar \pi}$$

• could have stuffed more than 1 wavelength in: 1, 3, 5, ...

$$\text{so } \Delta p \Delta x \geq \hbar \pi$$

• could have made wavelength smaller rather than larger.

• Why π and not $\frac{1}{2}$? This approximation too simple. But it gets the point

• Δp is independent of $p = \frac{2\pi\hbar}{\lambda}$

• Fill in wavelengths between $\frac{2\pi}{\lambda}$ and $\frac{2\pi}{\lambda \pm \Delta\lambda}$: that suppresses peaks way outside $\pm \Delta x$ (see spreadsheet).

Uncertainty + Bound States p. 341

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The uncertainty principle (or, that particles are waves) keeps the world from collapsing.

$$\hat{H} = \underbrace{\frac{p_x^2 + p_y^2 + p_z^2}{2m}}_{\substack{\text{kinetic energy} \\ \text{of } m = \text{electron mass} \\ \text{assuming proton } \infty \text{ mass}}} - \underbrace{\frac{e^2}{(\tilde{x}^2 + \tilde{y}^2 + \tilde{z}^2)^{1/2}}}_{\substack{\text{Coulomb} \\ \text{attraction;} \\ \tilde{x}, \tilde{y}, \tilde{z} \text{ are} \\ \text{coordinate of } e^- \\ \text{w/r to proton.}}}$$

$$\langle \hat{H} \rangle = \frac{\langle p_x^2 \rangle + \langle p_y^2 \rangle + \langle p_z^2 \rangle}{2m} - e^2 \left\langle \frac{1}{(\tilde{x}^2 + \tilde{y}^2 + \tilde{z}^2)^{1/2}} \right\rangle$$

seek to minimize $\langle \hat{H} \rangle$; that will be the ground state. We'll find that the closer you squish the electron into the proton, thereby reducing the potential energy contribution, the ensuing localization pushes the kinetic energy up (via uncertainty principle). Lowest energy is result of a compromise between the two terms.